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SYNTHESIS OF  $(2\underline{S}, 3\underline{R}, 4\underline{E}, 8\underline{E}) - \underline{N} - (2'\underline{R}) - 2' - HYDROXYHEXADECANOYL-9-$ METHYL-4,8-SPHINGADIENINE, THE CERAMIDE PORTION OF THE FRUITING-INDUCING CEREBROSIDE IN A BASIDIOMYCETE SCHIZOPHYLLUM COMMUNE

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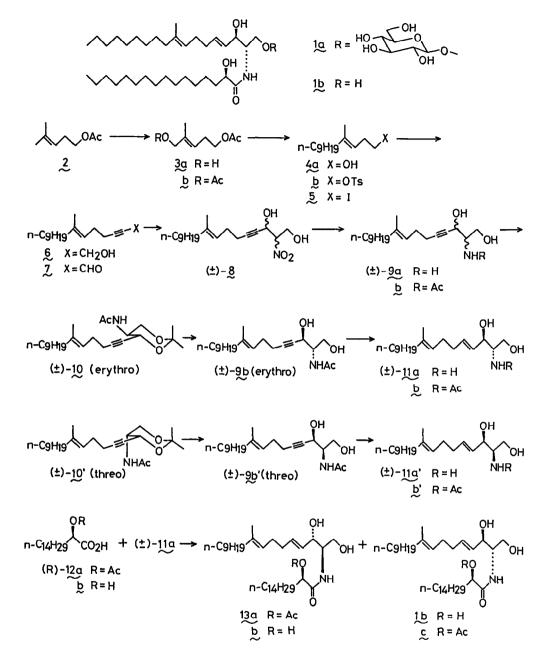
<u>Summary</u> : A total synthesis of the natural enantiomer of the title compound was accomplished, which confirmed the structure proposed for the fruiting-inducing cerebroside of <u>Schizophyllum</u> commune.

Fruiting body formation in <u>Basidiomycetes</u> is indeed a spectacular phenomenon especially to those who love to taste mushrooms. Its mechanism, however, is still a mystery in spite of the tremendous efforts to clarify it. Very recently Kawai and Ikeda found that the fruiting body formation of <u>Schizophyllum commune</u> (Japanese name : Suéhiro také) can be stimulated by some cerebrosides in its mycelia.<sup>2)</sup> They then identified one of the active substances as  $(2\underline{S}, 3\underline{R}, 4\underline{E}, 8\underline{E}) - \underline{N} - (2'\underline{R}) - 2' - hydroxyhexadecanoyl-1-\underline{O}-\beta-D-glucopyrano$ syl-9-methyl-4,8-sphingadienine <u>la</u>,<sup>3)</sup> which had previously been isolated froma sea anemone (<u>Metridium senile</u>) by Karlsson <u>et al</u>.<sup>4)</sup> Such a minute amount of<u>La</u> as 0.1 µg induced the fruiting body formation of <u>S.commune</u>, and thecorresponding ceramide <u>lb</u> lacking the sugar portion was also active.<sup>3)</sup>

This remarkable bioactivity of  $l_{e}$  prompted us to synthesize it so as to confirm the proposed structure.<sup>5)</sup> In this communication will be described a synthesis of the ceramide  $l_{e}$  with correct stereochemistry. Our synthetic  $l_{e}$  was highly active in inducing the fruiting body formation of <u>S.commune</u>.

Construction of the sphingadienine portion of 1b started from the known homoprenyl acetate 2.<sup>6)</sup> Oxidation of 2 with SeO<sub>2</sub> was followed by NaBH<sub>4</sub> reduction to give 3a, whose acetylation  $(Ac_2O/C_5H_5N)$  yielded 3b (38.0 % from 2). This was treated with <u>n-C<sub>8</sub>H<sub>17</sub>MgBr/THF</u> in the presence of Li<sub>2</sub>CuCl<sub>4</sub> to give 4a (94.2 %). The corresponding tosylate 4b was submitted to the Finkelstein reaction (NaI/acetone) to afford 5 (96.2 %). Alkylation of HC=CCH<sub>2</sub>OTHP with 5 was effected with <u>n</u>-BuLi in THF/HMPA (-15°, 1.5 hr; 0°, 1.5 hr), yielding 6 (44.3 %) after the removal of the THP group with <u>p</u>-TsOH/MeOH (room temp, 18 hr). The alcohol 6 gave an aldehyde  $\chi$  (92.4 %) upon treatment with MnO<sub>2</sub> in pet ether.

Condensation of  $\chi$  with 2-nitroethanol in the presence of  $K_2CO_3$  in MeOH<sup>7</sup>



afforded a nitro diol § as a diastereomeric mixture. Because of the instability of §, its chromatographic separation into the <u>erythro-</u> and <u>threo-</u>isomers was unsuccessful. Conversion of § into the corresponding acetonide<sup>8</sup>) was not successful either due to the retro-aldol reaction. The diastereomeric nitro diol § was therefore directly reduced with Zn/EtOH-conc HCl. The resulting §a was acetylated ( $Ac_2O/MeOH$ ) to give §b, whose EtOAc soln deposited a crystalline mass, mp 95.5%96.5% (24.6% from 7). This was later shown to be (±)-<u>threo-</u>\$b' (<u>vide infra</u>). The mother liquor was treated with Me<sub>2</sub>C(OMe)<sub>2</sub> and

PPTS in acetone to give a mixture of  $(\pm) - \frac{10}{40}$  and  $\frac{10}{20}$ . This was separated by medium pressure LC (SiO<sub>2</sub>) to give  $(\pm)-\underline{erythro}-\lambda$  (21.5% from 7) and  $(\pm)-\underline{threo}-\lambda$ 10' (4.7% yield from 7). The assignment of the relative configuration between C-2 and C-3 of 10 was made possible by <sup>1</sup>H-NMR spectroscopy according to our previous observation.<sup>8,9)</sup> The one with an equatorial -NHAc [ $\delta$  (CDCl<sub>3</sub>) 4.58(1H, d,J=7 Hz, -CHC=CCH<sub>2</sub>-)] was the erythro-isomer  $(\pm)$ -10, while the other with an axial-NHAc[δ(CDCl<sub>3</sub>) 4.80 (1H,d,J=2 Hz, -CHC≡CCH<sub>2</sub>-)] was the threo-isomer (±)-10'. The aforementioned crystalline  $(\pm) - \frac{1}{20}$  yielded  $(\pm) - \frac{10}{10}$  upon acetonide formation, manifesting its threo-stereochemistry. Removal of the acetonide protective group of (±)-10 was effected by treatment with <u>p</u>-TsOH/<u>i</u>-PrOH (40°, 40 min) to give (t)-erythro-9b. Prior to the reduction of the triple bond, (±)-erythro-9b was treated with KOH/aqMeOH (reflux, 6 hr) to remove the Nacetyl group. The resulting (±)-erythro-9a was reduced with LAH/THF to give (±)-erythro-lla (42.6% from 9b). Similarly (±)-threo-9b' yielded (±)-threo-11a' (52.7% from 9b'). These two were separately N-acetylated (Ac,O/MeOH) to give (±)-(4E,8E)-2,3-erythro-2-acetamino-9-methyl-4,8-octadecadiene-1,2-diol 11b and its three- isomer 11b'. The <sup>1</sup>H- and <sup>13</sup>C-NMR spectra of these two products were compared with those of the N-acetylnonadecasphingadienine derived from the natural cerebroside la of S. commune. A part of <sup>13</sup>C-NMR data as shown in Table 1 indicated that the sample derived from the natural product possesses erythro- stereochemistry. The 400 MHz <sup>1</sup>H-NMR spectrum of (±)-erythrollb was identical with that of the sample of the natural origin.<sup>10)</sup>

The remaining task was to prepare  $(\underline{R})$ -2-acetoxyhexadecanoic acid  $\frac{1}{2}$  and to N-acylate  $(\pm)$ - $\frac{1}{14}$  with  $(\underline{R})$ - $\frac{1}{12}$ . By employing  $(\underline{R})$ - $\frac{1}{12}$  as the acylating agent, we expected the separation of the resulting diastereomeric mixture in a later stage. The desired acid  $(\underline{R})$ - $\frac{1}{12}$  was prepared by the acetylation  $(Ac_2O/C_2H_5N)$  of the corresponding hydroxy acid  $(\underline{R})$ - $\frac{1}{12}$ , mp 92 $^{0}$ 93°;  $[\alpha]_D^{20}$ -2.9° (c= 1.03, CHCl<sub>3</sub>)[lit.<sup>12</sup>) mp 93.3 $^{0}$ 93.5°;  $[\alpha]_D^{-3.2°}$  (CHCl<sub>3</sub>)].<sup>11 $^{10}$ 14)</sup> This, in turn, was obtained by deaminating  $(\underline{R})$ -2-aminohexadecanoic acid, which was prepared by the enzymatic resolution (amino acylase)<sup>15</sup>) of the known racemate.<sup>16</sup>) Acylation of  $(\pm)$ - $\frac{1}{14}$  with  $(\underline{R})$ - $\frac{1}{12}$  in the presence of EtN=C=N(CH<sub>2</sub>)<sub>3</sub>NMe<sub>2</sub>·HCl<sup>14</sup>) gave a diastereomeric mixture of  $\frac{1}{3}$  and  $\frac{1}{3}$  (45.6% from  $\frac{1}{3}$ ). This was dissolved in CHCl<sub>3</sub> and treated with NaOH/MeOH (room temp, 1 hr) to give a mixture

Carbon (No.)	Natural	NaturalSynthetic (CDCl3, 22.6 MHz)(CDCl3, 100 MHz)(±)-erythro-llb (±)-threo-llb	
	(CDC1 <sub>3</sub> , 100 MHz)		
-CH <sub>2</sub> O-(C-1)	62.19	62.1	63.7
)CHN< (C-2)	54.36	54.5	54.9
)CHO- (C-3)	74.48	74.2	72.5

Table 1. <sup>13</sup>C-NMR data of the natural and synthetic 2-acetamino-9-methyl-4,8-octadecadiene-1,2-diol of 13b and 1b. Chromatographic separation (Merck Lobar column, Lichroprep $^{\textcircled{B}}$ Si 60 (40 $\circ$ 63 µm); Elution with CHCl<sub>3</sub>-MeOH=150:1) of the mixture gave 13b (22.3 \* from  $l_{a}$ ,  $[\alpha]_{D}^{21}+10.6^{\circ}$  (c=0.54, CHCl<sub>3</sub>), and  $l_{D}$ , mp 62 $\circ$ 64°;  $[\alpha]_{D}^{21}+6.4^{\circ}$  (c= 0.76, CHCl3). An authentic sample of 1b, prepared from 1a by the method of Hammarström<sup>17)</sup> showed mp 59 $\cdot$ 61°;  $[\alpha]_D^{21}$ +7.3±0.4° (c=0.25, CHCl<sub>3</sub>). The identity of the natural and synthetic 1b was confirmed by the comparison of IR, 1H-NMR (400 MHz), <sup>13</sup>C-NMR (25 MHz) and HPTLC using three different solvent systems.<sup>18)</sup> The final proof of the identity of our synthetic 1b with the ceramide 1b of natural origin was its very strong fruiting-inducing activity on S.commune. Indeed the specific activity of our lb (15,000 units/mg) was higher than that (10,000 units/mg) of the natural cerebroside la itself. The diastereomer  $\ensuremath{\mbox{l}3b}$ was less active (2,000 units/mg).<sup>19)</sup>

In conclusion, the structure la proposed for the fruiting-inducing cerebroside of Schizophyllum commune was confirmed by synthesizing lb.

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  10) <sup>1</sup>H-NMR data (400 MHz, CDCl<sub>3</sub>) of (±) -erythro-1Lb: 0.88 (3H,t,J=7.0Hz), 1.24 (12H,br.s), 1.36 (2H,m), 1.58 (3H,s), 1.94 (2H,t,J=8.0 Hz), 2.03 (3H,s), 2.08 (4H,br.s), 3.50 (1H,br.s), 3.58 (1H,br.s), 3.68 (1H,deformed d, J=11.0 Hz), 3.88 (1H,dt,J<sub>1</sub>=7.6 Hz, J<sub>2</sub>=3.8 Hz), 3.92 (1H,dd,J<sub>1</sub>=11.0 Hz, J<sub>2</sub>=3.8 Hz), 4.30 (1H,br.s), 5.09 (1H,t,J=6.0 Hz), 5.53 (1H,dd,J<sub>1</sub>=15.6 Hz, J<sub>2</sub>=6.4 Hz), 5.79 (1H.ddd,J<sub>1</sub>=15.6 Hz, J<sub>2</sub>=6.4 Hz), 4.55 (1H,d,J=7.0 6.4 Hz), 5.79 (lH,ddd,J<sub>1</sub>=15.6 Hz, J<sub>2</sub>=6.4 Hz, J<sub>3</sub>=5.5 Hz), 6.55 (lH,d,J=7.0 Hz).
- 11) (R)-2-Hydroxyhexadecanoic acid 12b was known as a constituent of wool wax. 12,13) Its (S)-enantiomer was synthesized by Horn et al.,13) while Hammar-ström reported the synthesis of (R)-12b (without experimental details).
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  18) Physical data of synthetic lb: <sup>1</sup>H-NMR (400 MHz, CDCl<sub>3</sub>) & 0.87 (6H,t,J=7.0 Hz), 1.20v1.45 (40H,m),1.58 (3H,s), 1.95 (2H,t,J=7.5 Hz), 2.08 (4H,br,s), 3.20 (1H,br.s), 3.55 (1H,br.s), 3.70 (1H,br,s),3.74 (1H,deformed d, J=11.0 Hz), 3.87 (1H,dd,J<sub>1</sub>=11.0 Hz, J<sub>2</sub>=4.0 Hz), 3.91 (1H,dt,J<sub>1</sub>=8.0 Hz, J<sub>2</sub>=4.0 Hz), 4.11 (1H,dd,J<sub>1</sub>=7.5 Hz, J<sub>2</sub>=3.5 Hz), 4.27 (1H,br,s), 5.09 (1H,t,J=6.0 Hz), 5.52 (1H,dd,J<sub>1</sub>=15.5 Hz, J<sub>2</sub>=6.5 Hz), 5.79 (1H,dt,J<sub>1</sub>=15.5 Hz, J<sub>2</sub>=6.0 Hz), 7.21 (1H,d,J=8.0 Hz); <sup>13</sup>C-NMR(25 MHz, CDCl<sub>3</sub>) & 14.1, 16.0, 22.7, 25.2, 27.6, 28.1, 29.4, 29.6, 29.7, 31.9, 32.6, 34.7, 39.7, 54.5, 61.9, 72.5, 74.1, 123.1, 128.6, 134.0, 136.2, 175.5; HPTLC (Merck Kieselgel 60F<sub>254</sub>) Rf 0.53 (CHCl<sub>3</sub>-MeOH=9:1), Rf 0.60 (<u>n</u>-hexane-acetone=1:1), Rf 0.16 (C<sub>6</sub>H<sub>6</sub>-EtOAc=1:4). EtOAc=1:4).
- 19) We thank Mr.G.Kawai of Noda Institute for Scientific Research, Noda, Chiba, for his kind gift of la and for the bioassay of our synthetic materials.